Stereoselective Synthesis of Polyfunctional Di- and Trisuhstituted Cyclopentane Derivatives using a New Palladium Catalyzed Cyclization.

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Abstract : Various functionalized 5-hexenyl iodides undergo a radical ring closure to give cyclopentylmethylzinc iodides when treated with Et₂Zn in the presence of PdCl₂(dppf) (2 mol %). In the case of 4-substituted 5-hexenyl iodides trans-cyclopentanes are obtained (> 99 . 1) whereas 1-substituted-5-hexenyl iodides provide cis-cyclopentanes (ca. 80 : 20). The presence of a substituent at both positions considerably improves the *cis* stereoselectivity up to 95 : 5.

I : **1) EtzZn (2 cqmv.), PdClz(dppf) (2 mol%), 20 'C, 5-20h 2) CuCM2L1Cl3) ethyl 2-(bromometbyl)acryIate**

The stereoselective preparation of di- and trisubstituted cyclopentanes is an important synthetic goal since such ring units are present in numerous natural products of biological interest.² Recently, we have shown that primary alkyl iodides and Et2Zn undergo an efficient iodine-zinc exchange reaction in the presence of catalytic amounts of Pd(II) or Ni(II) salts. 5-Hexenyl iodides are rapidly cyclized (25 °C, 2h) under these conditions providing cyclopentylmethylzinc iodides.³ Although a wide range of both anionic⁴ and radical⁵ ring closures are known, the present method combines in a unique way, advantages of both cyclization modes. The ring closure, which we believe is of radical nature, 3 produces an organometallic species which can subsequently be trapped by a variety of electrophiles. Contrary to most anionic cyclizations, a large number of functionalities can be present in the unsaturated iodide. We wish now to report some remarkable stereochemical aspects of the ring closure involving secondary iodides as substrates and leading stereoselecttvely to functionalized di- and trisubstituted cyclopentane derivatives. We have found that 4 substituted-5-hexenyl iodides such as la-b undergo a stereospecific ring closure (Et₂Zn(2 equiv.); PdCl₂(dppf)⁶(1.5 mol %) leading, after a transmetalation with CuCN·2LiCl⁷ and allylation with ethyl α -(bromomethyl)acrylate, ⁸ to *trans*-1,2-disubstituted cyclopentane derivatives **2a-b** in 41-67 % yield (trans : cis > 99 : 1; eq 1).⁹ In strong contrast, *secondary* iodides¹⁰ such as **3a-f** (substituent at position 1) afford preferentially, after transmetalation and quenching with an electrophile, the cis-1,2-disubstituted cyclopentanes **4a-41** m 70 : 30 to 86 : 14 cis *: truns* ratios (eq. 2 and Table 1). The cis-selectivity of the nng closure increases when the size of the substituent R at position 1 decreases (compare entries l-3 of Table I). The cyclized zinc reagent can be trapped after transmetalation with various electrophiles (3-10do-2cyclohexenone¹¹ (entries 1-3), ethyl α -(bromomethyl)acrylate⁷ (entries 4, 6 and 8), 2-cyclohexenone (entry 5), cis -1-acetoxy-4-chloro-2-cyclohexene¹² (entry 7) and ethyl propiolate (entries 9 and 10) in satisfactory yields (52-87 %). This new cyclization has some remarkable features. The substrates 3 tolerate the presence of various functionaliues (see **3d-3f').** Especially interesting is the nng closure of 3e which bears a good leaving group in β -position (OPiv). If this iodide is converted during the cyclization process into an organozinc derivative, an immediate elimination which would lead to 1,6-heptadiene, is expected to occur. However, this ring closure proceeds very smoothly showing unambiguously that no organozmc species is formed pnor to ring closure. The method allows the performance of tandem-cychzation reactrons. Thus, the diemc iodide **3f gives** after transmetalaaon with CuCN-2LiCl and addition to ethyl propiolate, the bicyclic unsaturated ester **4k** (cis-ring junction, > 98 % E; exo : endo : 57 : 43) and the mono-cyclic *truns* dienic ester 4j (the rrms stereochemistry hampers the second ring closure); the ratio between $4k$ and $4j$ is $86:14$.

The cis-stereoselectivity can be dramatically improved by using an unsaturated iodide which bears a substttuent at position 1 *and* at posttion 4. Thus the ring **closure** of 5 (used as a 1 : 1 mixture of diastereoisomers) produces after allylation the trisubstituted cyclopentane derivative 6 with high stereoselectivity *(trans : cis* between C-1 and C-2 (> 99 : 1); *trans : cis* between C-2 and C-3 (5 : 95); eq. 3). Further information about the mechanism of the reaction is obtained from the reaction of exo and endo-7-10dobicyclo $[2.1.0]$ heptane 7 with Et₂Zn (ca. 2 equiv., 5 mol % PdCl₂(dppf)) in THF. Both of these iodides are converted to the same exo-organozinc reagent which, after the usual allylation reaction, provides in a stereoconvergent manner the exo-substituted bicyclic adduct 8 in 60 % yield (endo: exo 4: 96; eq. 4).¹³ This stereoconvergence indicates that the insertion of palladium proceeds through a radical mechanism.¹⁴ Compared to conventional radical cyclizations, better regioselectivities and similar stereoselectivities are observed. Thus I-(3-butenyl)-2-iodocyclohexane 9 provides, after allylaaon, the bicyclic esters **10** as the only products (exo : endo 20 : 80; 90 % yield). Similar or lower stereoselectivities are found in Bu3SnH induced radical cyclizations; furthermore in such cases, the reaction is complicated by the presence of non-cyclized products resulting in significantly lower yields 15 (eq. 5).

1 1) Et₂Zn (2 equiv.), PdCl₂(dppf) (2 mol%), 20 °C, 5-20h 2) CuCN.²LiCl 3) ethyl 2-(bromomethyl)acrylate

In summary, 17 we have shown that various unsaturated secondary iodides undergo a unique radical cyclization in the presence of Et₂Zn and a catalytic amount of Pd(II) or Ni(II) salts providing an organometalhc species which can be trapped by various electrophiles. In most cases ca. 80 : 20 CIS : *truns* ratios are obtained. We have demonstrated that this cis selectivity can be considerably improved by mtroducing another substituent at position 4 allowing a highly stereoselecave synthesis of polyfunctional

Entry	Iodide of Type 3	Electrophile	Product of Type 4	cis:trans	Yield ^a
$\mathbf{1}$ $\mathbf 2$ 3 4	$3a: R = Me$ $3b: R = Et$ $3c: R = c-Hex$ ℐ CH ₂ CH ₃	O CO ₂ Et Br	50 ₅ R $4a: R = Me$ $4b: R = Et$ $4c: R = c$ -Hex CO ₂ Et CH ₃ 4d	78:22 75:25 70:30 75:25	$(\%)$ 62 81 65 87
5	3 _b c-Hex 3c		O c-Hex 4e CO ₂ Et	70:30 ^b	52
$\boldsymbol{6}$	(CH ₂) ₄ OAc 3d	CO ₂ Et ,Br QAc	OAc 4f ACQ	77:23	67
$\overline{7}$	(CH ₂) ₄ OAc 3d	ĊІ	4g OAc CO ₂ Et	77:23b	52
$\bf 8$	OPiv 3 _e	CO ₂ Et Br	OPiv 4 _h	78:22	87
9	OPIV 3e	HCEC-CO ₂ Et	CO ₂ Et OPIV 4i	78:22 ^c	80
10	3f	HC=C-CO ₂ Et	CO ₂ Et 4j CO ₂ Et 4k	86:14c,d	82

Table I. Polyfunctional Cyclopentanes Obtained by the Pd Catalyzed Cyclization.¹⁶

a All yields refer to analytically pure products; ^b Mixture of diastereoisomers; ^c >98% E isomer; ^d 4k is obtained as a 43:57 mixture of endo exo isomers;

trisubstituted cyclopentane derivatives (eq. 3). Extensions of this method and applications to the preparation of more complex polycyclic molecules are currently underway in our laboratories.

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- Y This high trans-sefectivity is in strong contrast with the uncatatyzed carbozincation of 3-alkoxy-I-iodo-5-hexene derivatives which gives in most cases 2 : 3 mixtures of cis and trans 1,2-disubstituted cyclopentanes (see ref. 41).
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- 16 The relative stereochemistry of the *cis -* cyclopentane derivatives has been assigned by NOESY-NMR- experiments on several products (2a, 4f). The exo- endo ratios of 8 and 10 have been determined respectively by iodolysts of the zinc intermediates and by comparison with the ¹³C NMR spectra of related compounds. Thus the minor diastereoisomer of 10 shows a down-field shift for most signals (ref. 15c).
- 17 Typical Procedure: Preparation of (E)-1-(3-carboethoxy-2-propen-1-yl)-2-pivaloyloxymethylcyclopentane 4j: A 3-necked flask equipped with a magneuc stirring bar, a thermometer and a gas mlet and charged with PdCl₂(dppf) (0.07 g, 2 mol %) in THF (5 mL) was cooled to -78 °C. 2-Iodo-6-heptenyl 2,2-dimethylpropanoate 3e (1.62 g, 5 mmol) and Et2Zn (1.0 mL, 1.23 g, 10 mmol) were added. After warming to 25 °C and stirring for 2 h, the solvent and excess Et₂Zn were removed (0.1 mm Hg, 25 °C, 1 h). After adding THF (5 mL) and cooling to -40 °C, CuCN 2LiCl (5 mmol) in THF (5 mL) was added, and the reaction **mixture was warmed to 0 'T** (5 mm) and cooled to -78 "C. Ethyl propiolate (0.39 g, 4.0 mmol) m THF (3 mL) was added and the reaction mixture was slowly warmed to -10 $^{\circ}$ C and stirred for 5 h. After the usual workup the residual oil was purified by flash column chromatography (ether : hexanes 1 : 9) to yield 4*j* as a clear oil (1.18 g, 80 %).